

Modeling Studies in Tropospheric Ozone Changes:

The Effect of Natural and Anthropogenic Sources and Climate Change

Cynthia Atherton
D. Bergmann
P. Connell
D. Rotman

LLNL, Atmospheric Science Division

Feb. 1998

Our work

thus far

Previously LLNL studied 3D global O₃ with a tropo-spheric model:

- GRANTOUR
- Fully predictive troposphere; specified stratosphere
- Meteorology from GCM
- Chemistry included oxidation of:

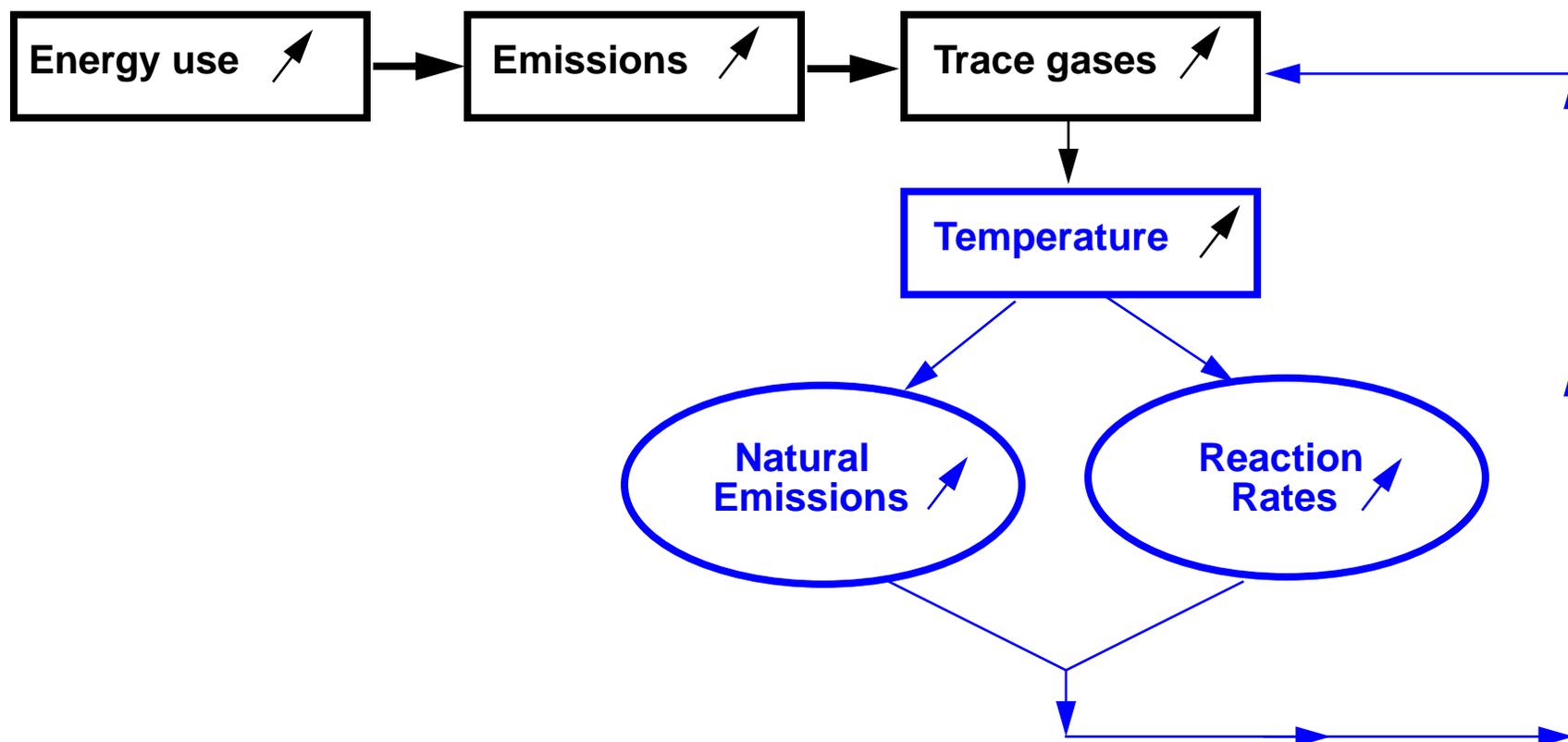
CO, CH₄, isoprene, ethane, propane, butane, ethene, propene, butene,
NO, NO₂, PAN, HNO₃,

O₃, OH, H₂O₂, HCHO, etc.

- Regional to global issues

Recent results

Use Global and Box Models to **understand**:



IPCC 1995: Temp increases 1 - 4.5K by 2100:

IPCC 1995: Temp increases 1 - 4.5K by 2100:

(Insert figure from IPCC here)

Annual mean temp. change (K) under Scenario IS92a, from 1795 to 2030, from CO₂ and aerosols (Mitchell et al., J. Climate, 10, 2365-2386, 1995)

Temperature increases will affect:

Reaction rates: $k = A \times \exp(-E_a/RT)$

Temperature dependent source emission rates:

Soil NO_x

Isoprene

Lightning NO_x

Soil NO_x emissions depend strongly on temperature

For example, for grasslands:

$$\log_{10} (NO \text{ flux}) = (0.0049 \pm 0.0008)(^{\circ} C^{-1})T_A - (0.83 \pm 0.13)$$

[] = ng N/m²-s

T_A = Air temperature

(Williams et al., **JGR 92**, 2,173 - 2,179, 1987)

$\Delta T = 2K \Rightarrow NO \text{ flux increases } 20\%$

$\Delta T = 4K \Rightarrow NO \text{ flux increases } 35 - 40\%$

Isoprene emissions increase exponentially with temperature

$$I(T, L) = I_s \cdot C_L \cdot C_T$$

$$C_L = \frac{\alpha c_{L1} L}{\sqrt{1 + \alpha^2 L^2}}$$

$$C_T = \frac{\exp\left[\frac{C_{T1} (T - T_s)}{RT_s T}\right]}{1 + \exp\left[\frac{C_{T2} (T - T_M)}{RT_s T}\right]}$$

I_s = Emission rate at T_s and PAR = 1000 $\mu\text{mol}/\text{m}^2\text{-s}$

L = Photosynthetically Active Radiation (PAR)

$T_M, \alpha, C_{L1}, C_{T1}, C_{T2}$ are specified from measurements

(Guenther et al., **JGR 98**, 12,609 -12,617, 1993)

$\Delta T = 2\text{K} \Rightarrow I(T, L)$ increases 20%

$\Delta T = 4\text{K} \Rightarrow I(T, L)$ increases ~40%

As the climate warms, lightning frequency increases:

$2 \times \text{CO}_2$ ($\sim 4.2^\circ\text{C}$) \Rightarrow NO_x from lightning increases:

13% over oceans

70% over land

(Price and Rind, *JGR* **99**, 10,823-10,831, 1994)

Result - chemical implications of climate change:

Some natural emissions increase as temperature increases

Temperature change	Soil NO _x	Isoprene	Lightning
$\Delta T = 2K$	+ 20%	+ 20%	-----
$\Delta T = 4K$	+ 40%	+ 40%	+70% (land) +13% (ocean)

Method:

Conduct baseline present-day simulation with global model.

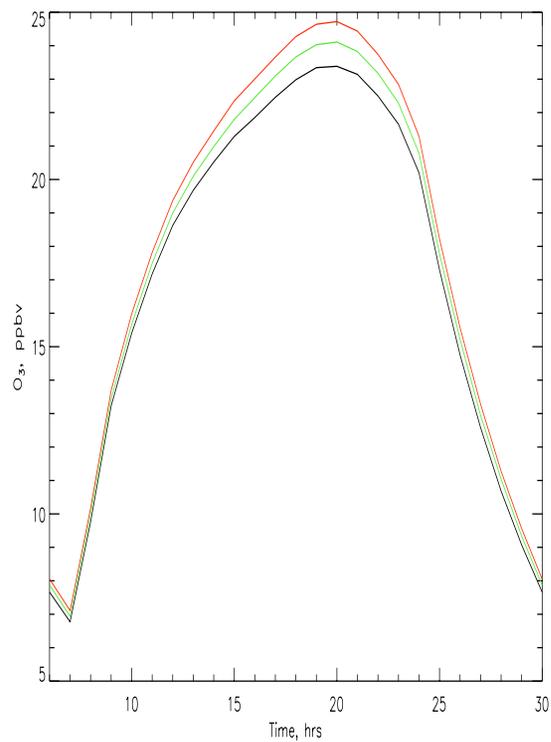
Extract concentrations and emissions for different regions.

Use box model to simulate regions for 3 scenarios.

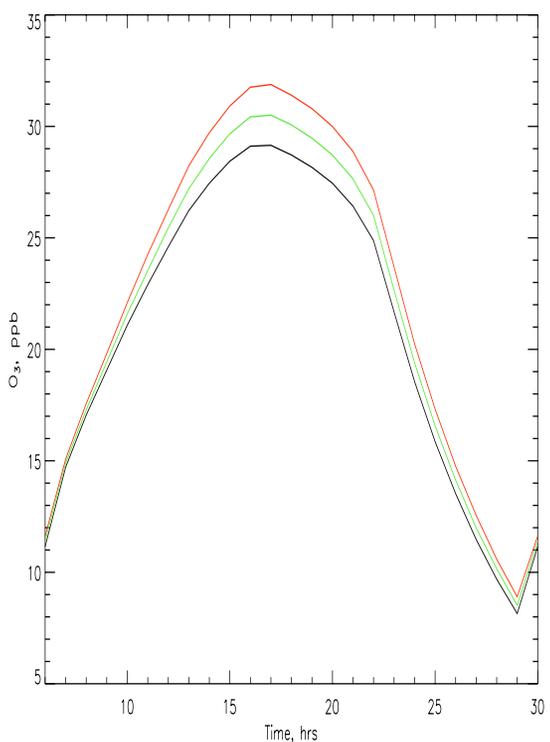
Scenario	Temperature	Soil NO _x	Isoprene	Lightning NO _x
Base	Base	Base	Base	Base
T+2K	$\Delta T = 2K$	+ 20%	+ 20%	-----
T+4K	$\Delta T = 4K$	+ 40%	+ 40%	+70% (land) +13% (ocean)

Changes in O₃ vary with location for Base, +2K, and +4K

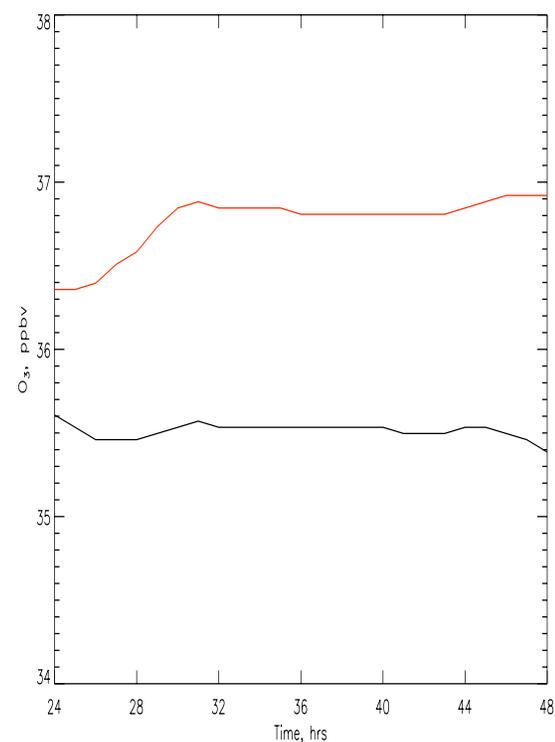
(42, -97.5)
surface



(-10, -70)
surface

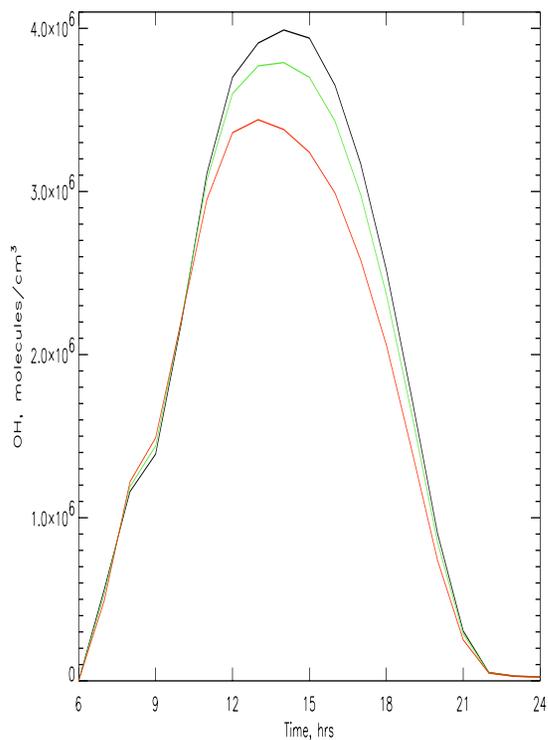


(24.4, 97.5)
sigma=0.664

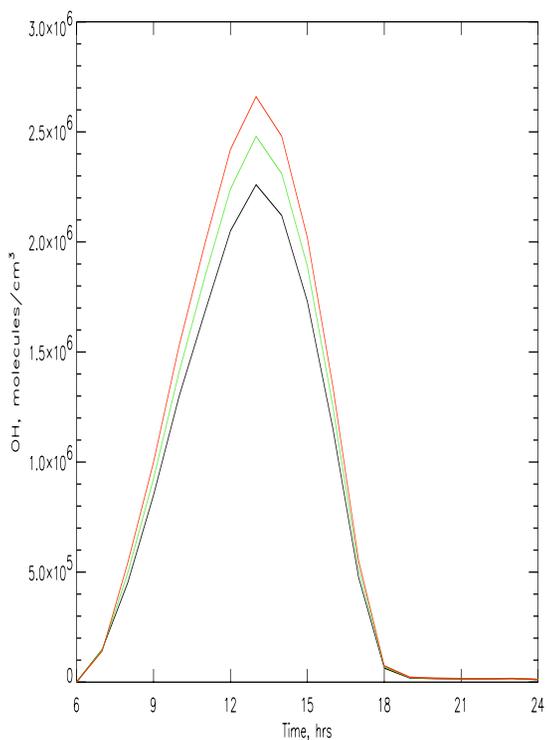


The OH may *increase* or *decrease* for Base, *+2K*, and *+4K*

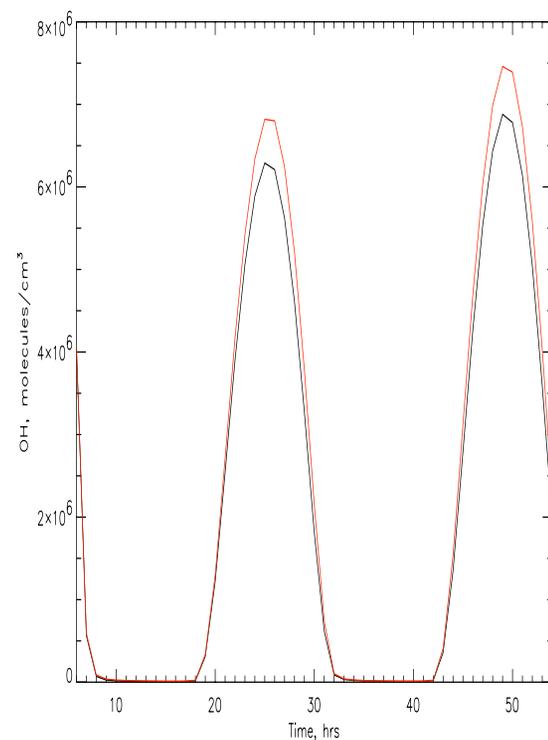
(42, -97.5)
surface



(-10, -70)
surface

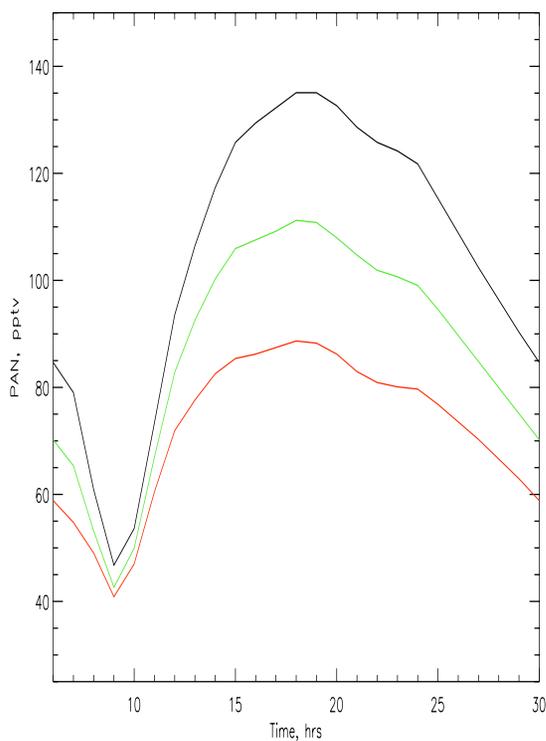


(24.4, 97.5)
sigma = 0.664

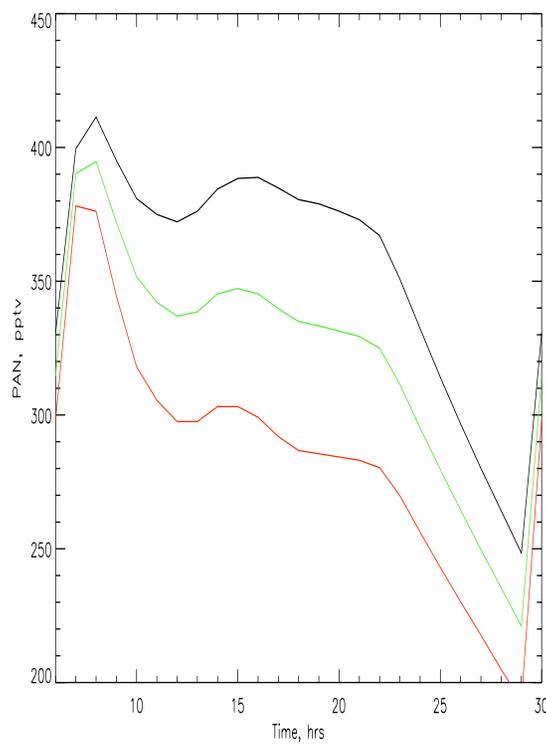


PAN concentrations change for Base, +2K, and +4K

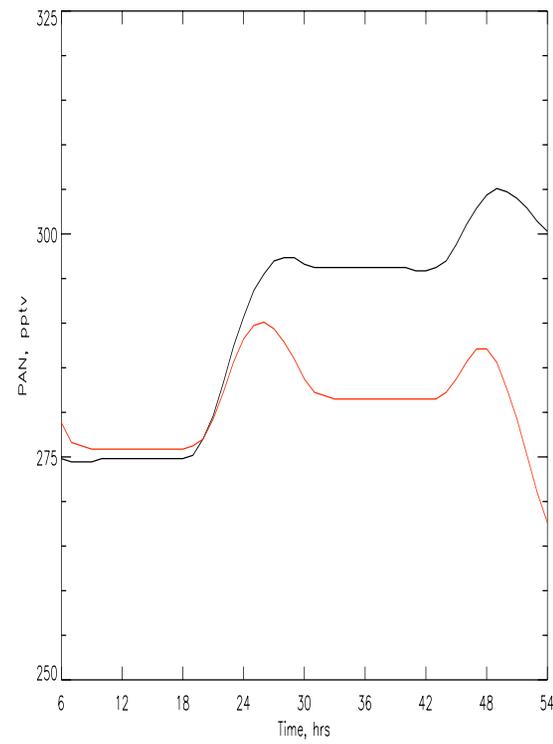
(42, -97.5)
surface



(-10, -70)
surface



(24.4, 97.5)
sigma = 0.664



Can we isolate the effects?

So far, have combined

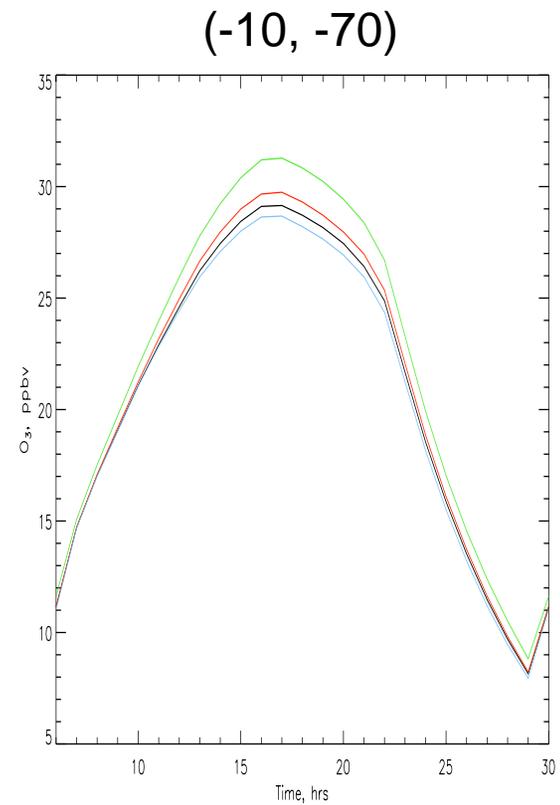
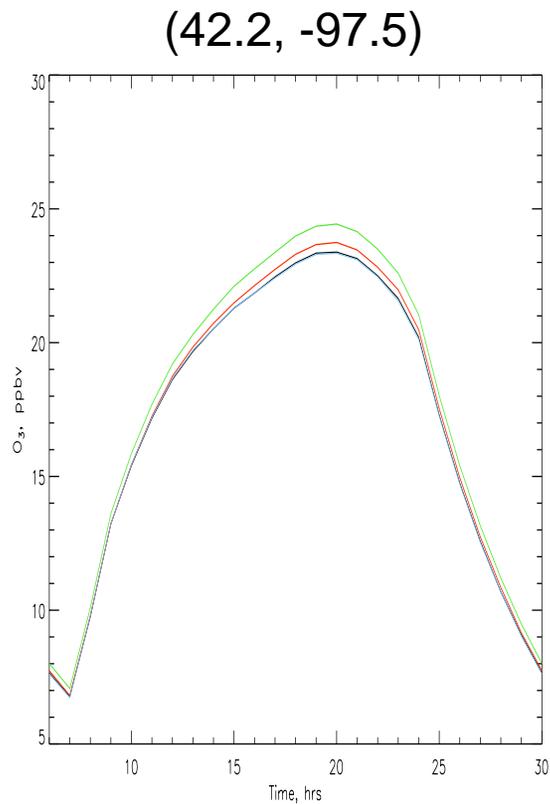
Temp increase + isoprene + lightning NO_x + soil NO_x increases

For surface locations examine four separate scenarios:

- (1) Base
- (2) Temperature + 4K only
- (3) Soil NO_x + 40% only
- (4) Isoprene + 40% only

What increase causes the greatest change in surface O₃?

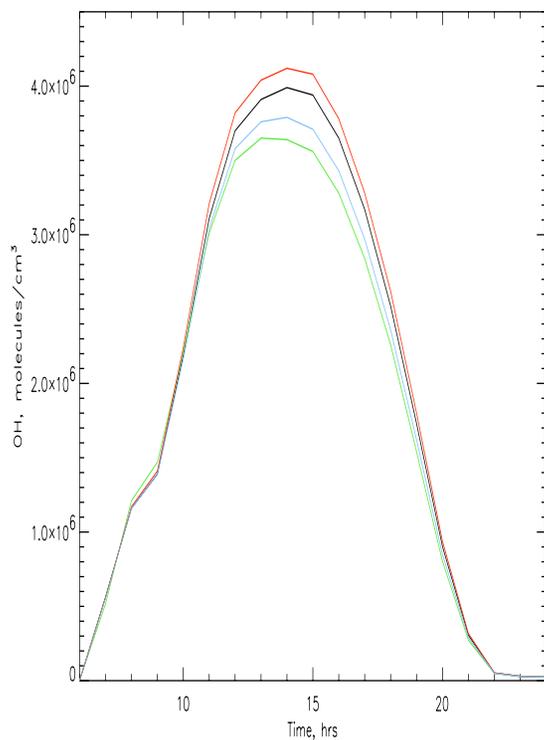
— Base — Temp+4K — Soil NO_x only — Isoprene only



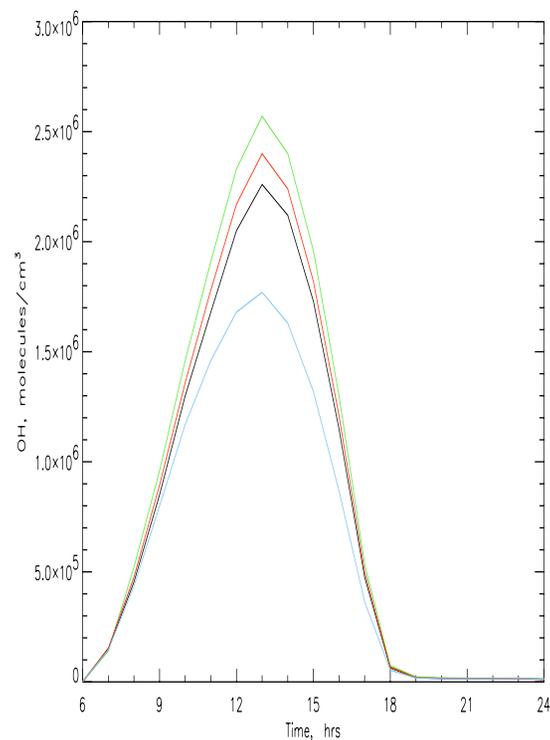
What increase causes the greatest change in surface OH?

— Base — Temp+4K — Soil NO_x only — Isoprene only

(42,2, -97.5)



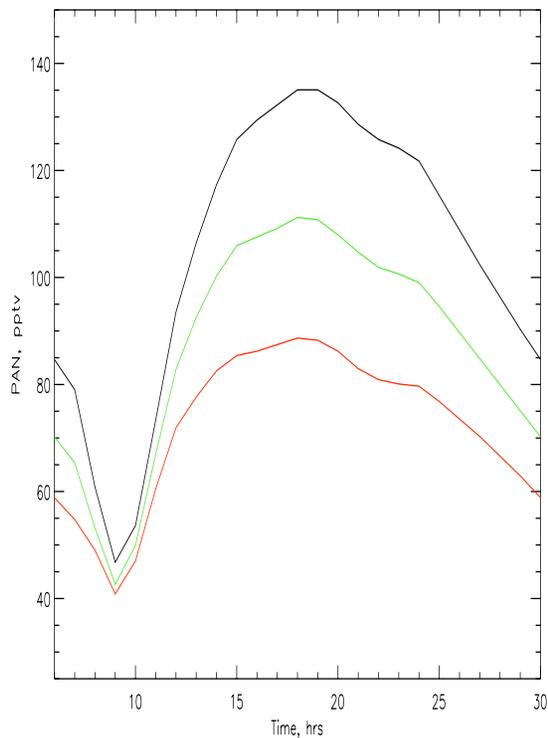
(-10, -70)



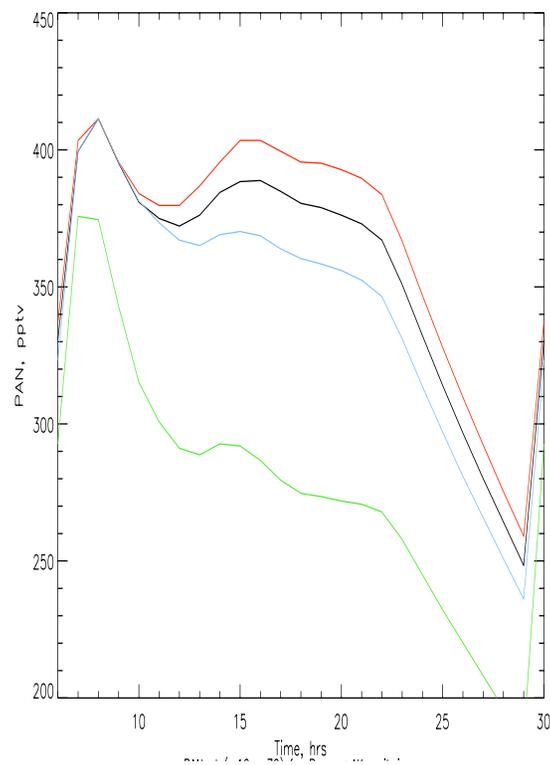
What increase causes the greatest change in surface PAN?

— Base — Temp+4K — Soil NO_x only — Isoprene only

(42.2, -97.5)



(-10, -70)



Result - chemical implications of climate change:

Some natural emissions increase as temperature increases

Temperature change	Soil NO _x	Isoprene	Lightning
$\Delta T = 2K$	+ 20%	+ 20%	-----
$\Delta T = 4K$	+ 40%	+ 40%	+70% (land) +13% (ocean)

Climate/chemistry effects are regional:

(42, -98) surface

O₃ ↑ 2 - 3 ppbv

OH ↓ ~ 20%

PAN ↓ 45 pptv

(-10, -70) surface

O₃ ↑ 3 - 4 ppbv

OH ↑ ~ 20%

PAN ↓ ~ 90 pptv

(25, 98) elevated (sigma=0.664)

O₃ ↑ > 2 ppbv

OH ↑ ~ 10%

PAN ↓ 30 pptv

Conc = COMPLEX fcn(T, isoprene, soil NO_x, lightning No_x)
(Note that PAN changes will affect far-field distributions, also.)

Future

work

LLNL is planning application studies with **IMPACT**:

- 3D, global
- Prognostic **troposphere AND stratosphere**
- High resolution meteorology from:

Assimilated wind fields **OR** GCM

2° latitude × 2.5° longitude

46 vertical layers (to 65 km)

Updated every 6 hours

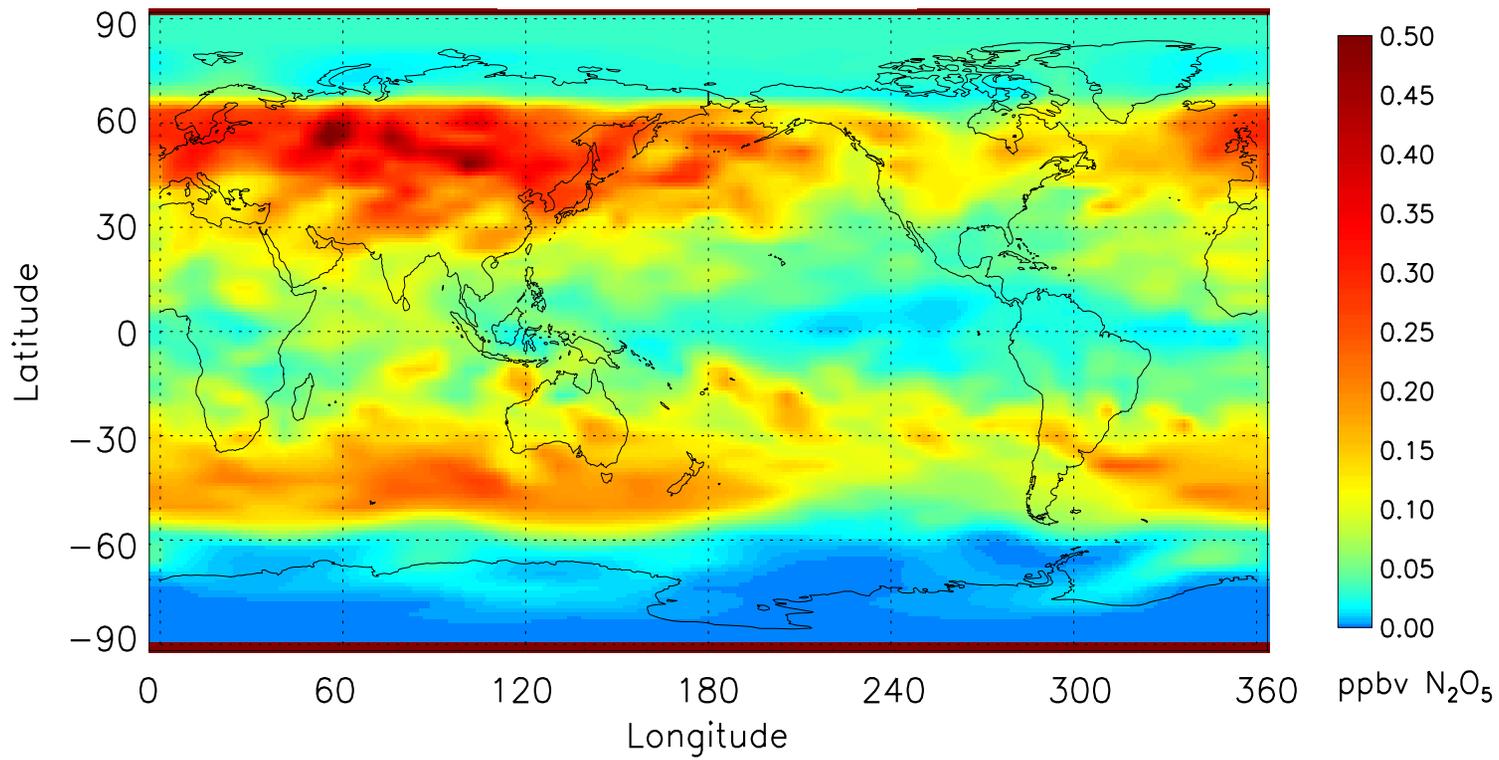
- Contains fully prognostic stratosphere.
- Currently adding tropospheric chemistry
- Massively parallel (computer) code

Our proposed work includes:

- Adding heterogeneous reactions of NO_3 and N_2O_5 on aerosols
- Evaluating the role of natural versus anthropogenic sources of O_3
- Examining HC vs. NO_x limitations (regionally, seasonally)
- Preliminary assessment of effective mitigation strategies

----->Comparison with observations throughout<-----

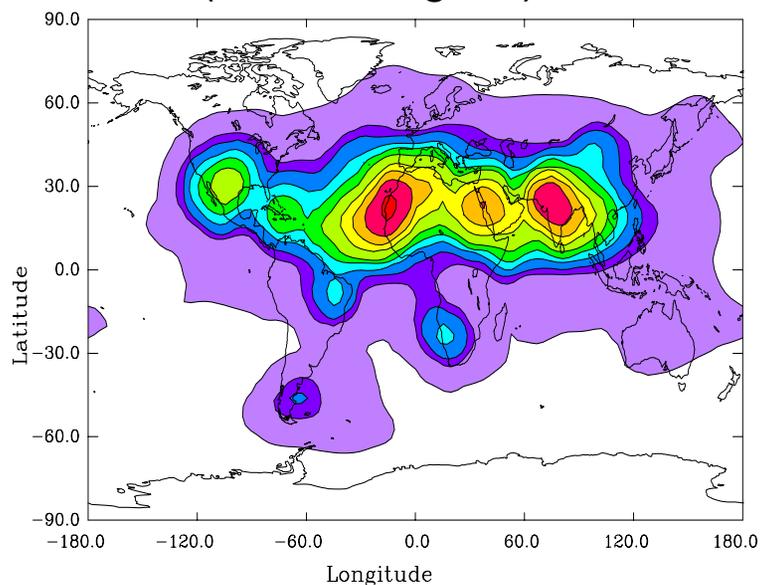
Below is the IMPACT simulated concentration of N_2O_5 at 55 hPa



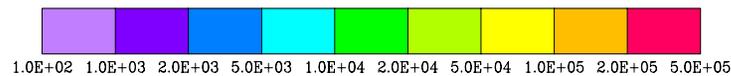
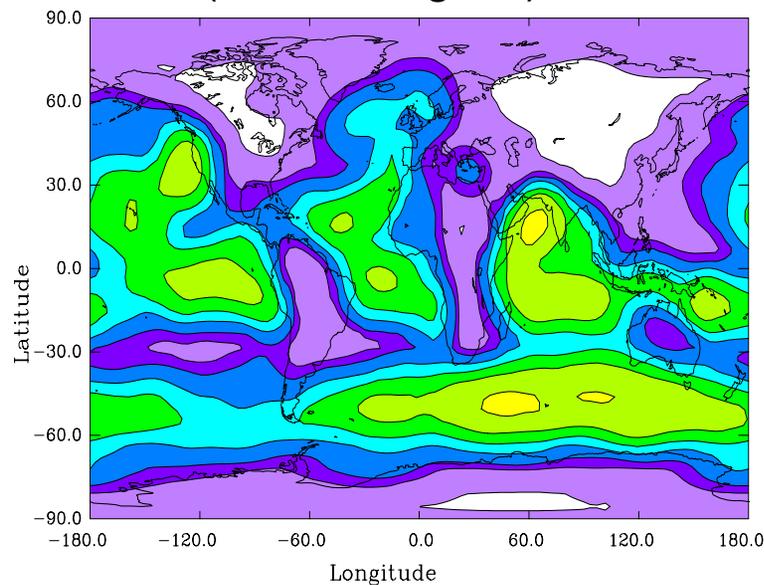
N_2O_5 , ppbv, 3 August, midnight at Greenwich

Incorporating heterogeneous reactions of NO_3 and N_2O_5 on aerosols will affect chemical cycles (seasonally and regionally):

July mineral aerosol
(surface, ng/m^3)



July seasalt aerosol
(surface, ng/m^3)



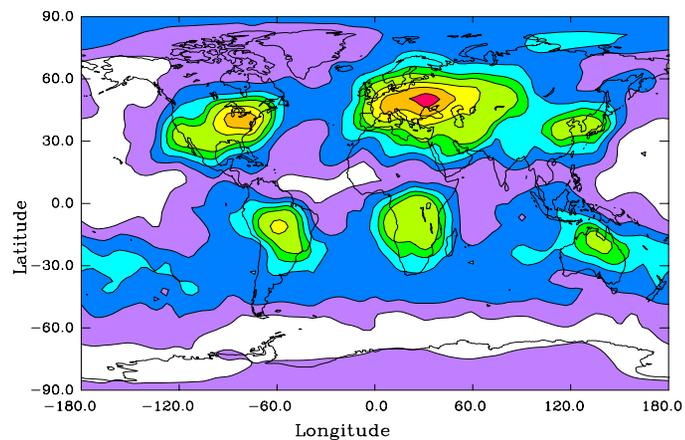
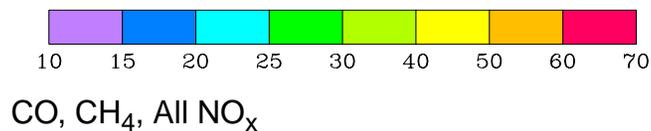
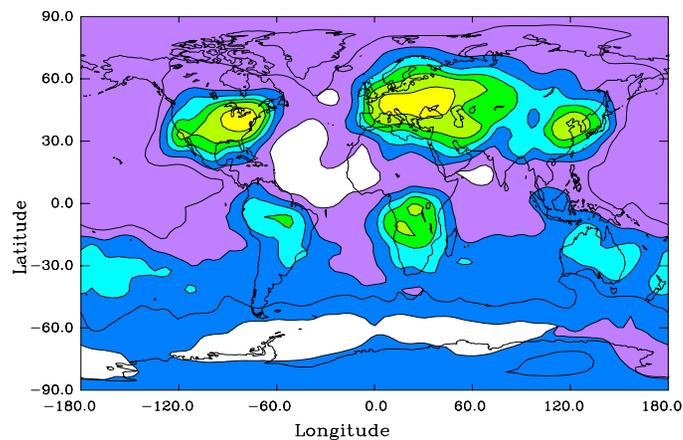
Also: sulfate, carbonaceous aerosols

IMPACT will quantify natural vs. energy-related sources of O₃ (regional and seasonal variations):

Natural sources

- a) Natural HCs (isoprene, terpenes, CH₄, etc.) and CO
- b) NO_x from lightning, soils

Preliminary GRANTOUR results - perpetual July - isoprene effect ONLY



IMPACT HC and/or NO_x sensitivity studies will address:

- Issues on a regional and seasonal basis
- Regions of HC and/or NO_x limitations
- Contribution of natural vs. energy-related emissions
- Effectiveness (O₃, etc.) of possible mitigation strategies

LLNL will further develop and apply global 3D models to help understand the role and importance of regional energy-related emissions on our atmospheric quality.

By understanding the role that both natural and anthropogenic emissions have, we can better develop energy-use and emission strategies to take us into the next century.