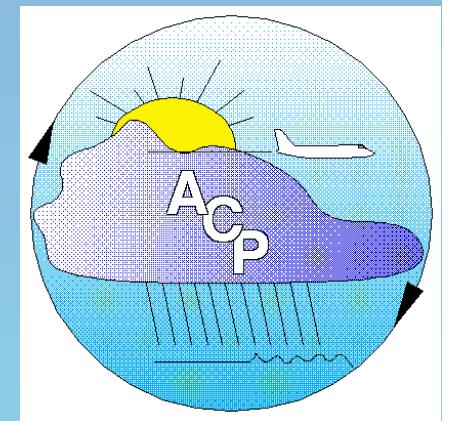


Chemistry and Micrometeorology of the Troposphere: Chemical and Micrometeorological Aerosol Model

**Carmen M. Benkovitz, Stephen E. Schwartz,
Robert L. McGraw, and Douglas L. Wright**
Environmental Chemistry Division
Department of Applied Science
Brookhaven National Laboratory
Upton, NY 11973



Background

★ HYPOTHESIS

#Anthropogenic emissions of SO₂ influence cloud and clear-sky albedo and can thus influence climate.

★ MECHANISMS

#Direct: enhancement of clear-sky albedo.

#Indirect: enhancement of average cloud albedo .

★ PROPERTIES

#Both forcing and results are highly variable in time and space.

★ EXISTING CONDITIONS

#Few observational data are available.

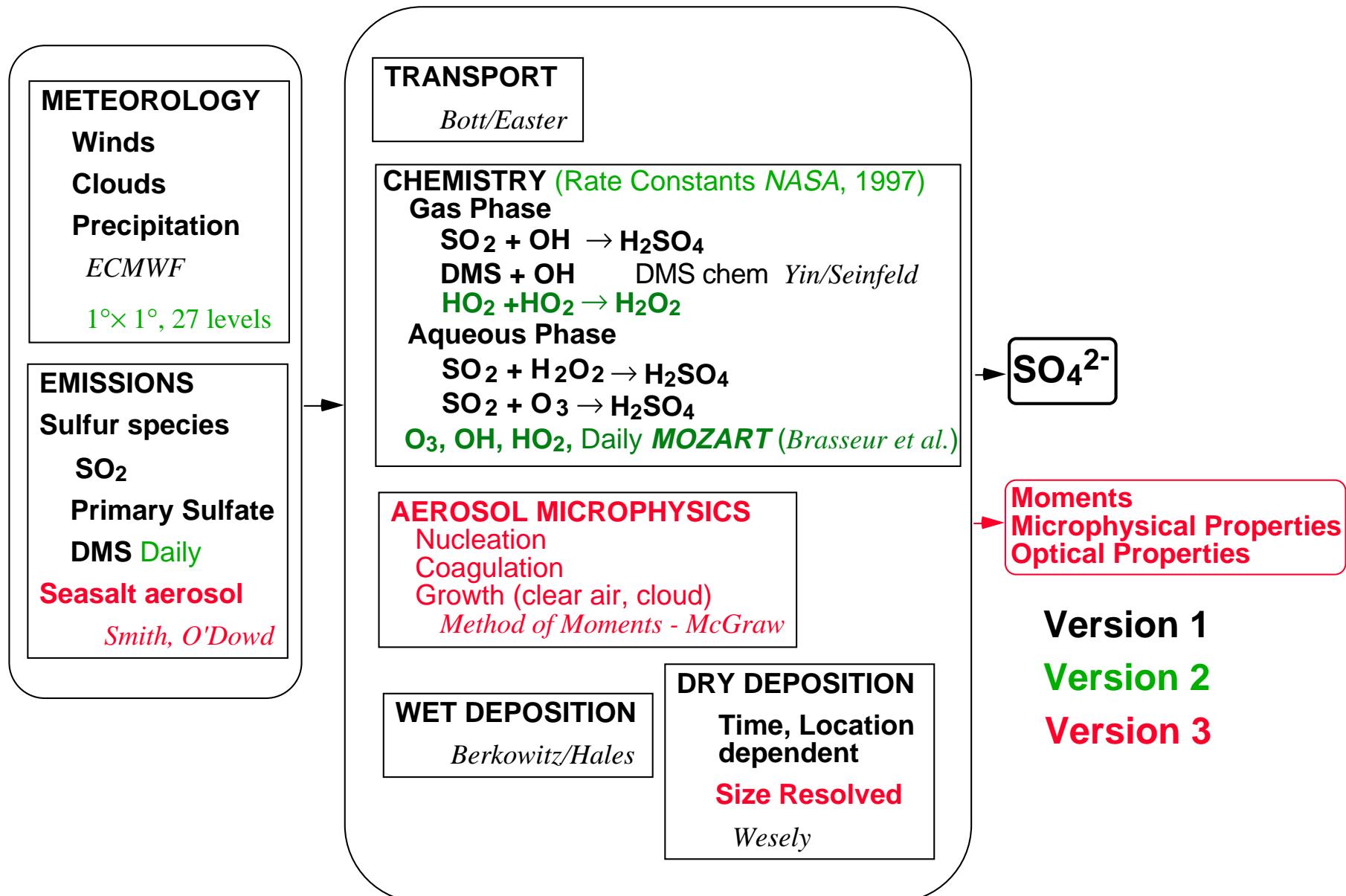
#Consistent estimates need modeling studies.

★ METHOD OF STUDY

#Model sulfate concentrations explicitly with high space and time resolution.

Aerosol Chemical Transport Model GChM-O

Global Chemistry Model Driven by Observation-Derived Meteorological Data



GChM-O Version 1

Specifications

★ Meteorology

#ECMWF uninitialized analyses.

1.125° x 1.125° horizontal resolution.

#15 vertical levels.

★ Transport

#Bott advection scheme as modified by Easter.

★ Dry deposition

#Time and location dependent using the methodology of Wesely and the ECMWF meteorological data.

★ Wet deposition:

Berkowitz/Hales.

GChM-O Version 1

Specifications

★ Chemistry:

#Gas-phase

- ! $\text{SO}_2 + \text{OH} \rightarrow \text{Sulfate}$
- ! $\text{DMS} + \text{OH} \rightarrow \text{Sulfate, MSA}$ (Yin/Seinfeld mechanism)

#Aqueous-phase (precipitating clouds only)

- ! $\text{SO}_2 + \text{H}_2\text{O}_2 \rightarrow \text{Sulfate}$
- ! $\text{SO}_2 + \text{O}_3 \rightarrow \text{Sulfate}$
- ! Cloud water pH = 4.5

#Oxidants

- ! [OH] from Spivakovsky et al.
- ! Seasonal $[\text{H}_2\text{O}_2]$ and $[\text{O}_3]$

#Rate constants from NASA (1991).

Species Represented by the Model

Source ↓	SO ₂		Sulfate			DMS	MSA	
	Process →	Primary	Secondary Gas Phase	Primary	Secondary Gas Phase	Secondary Aqueous Phase	Primary	Secondary Gas Phase
Anthropogenic 140°W to 30°W	X			X	X	X		
Anthropogenic 30°W to 60°E	X			X	X	X		
Biogenic		X			X	X	X	X
External		X			X	X		

Primary = directly emitted

Secondary = formed by atmospheric chemistry

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Simulations

★ Temporal domain

#June 28 to July 31, 1986.

#October 14 to November 15, 1986.

#January 28 to February 28, 1987.

#March 28 to April 30, 1987.

★ Geographic domain

#140°W to 68°E.

#12°N to 81°N.

★ Emissions

#Anthropogenic SO₂ and primary sulfate

! NAPAP for North America

! EMEP for Europe

! Dignon (1989) for all other areas

#Biogenic DMS from Bates et al. (1992); ocean DMS distributed using Coastal Zone Color Scanner (CZCS).

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Analysis of Model Input and Results

★ Meterology

- # Wind vector plots.
- # Synoptic charts.

★ Comparison with observations

- # Time series.
- # Histograms.

★ Graphics

- # Maps of sulfate column burdens.
- # Time series of relative contribution of sources.

★ Autocorrelation studies

- # $1/e$ decay of temporal and spatial autocorrelation.

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Analysis of Model Input and Results

★ Animations

#Animations of the development of the sulfate column burden.

#Web site: http://www.ecd.bnl.gov/sulfate_model.html

★ Analysis of meteorological influences on sulfate column burdens.

★ References

#Benkovitz et al., JGR 1994.

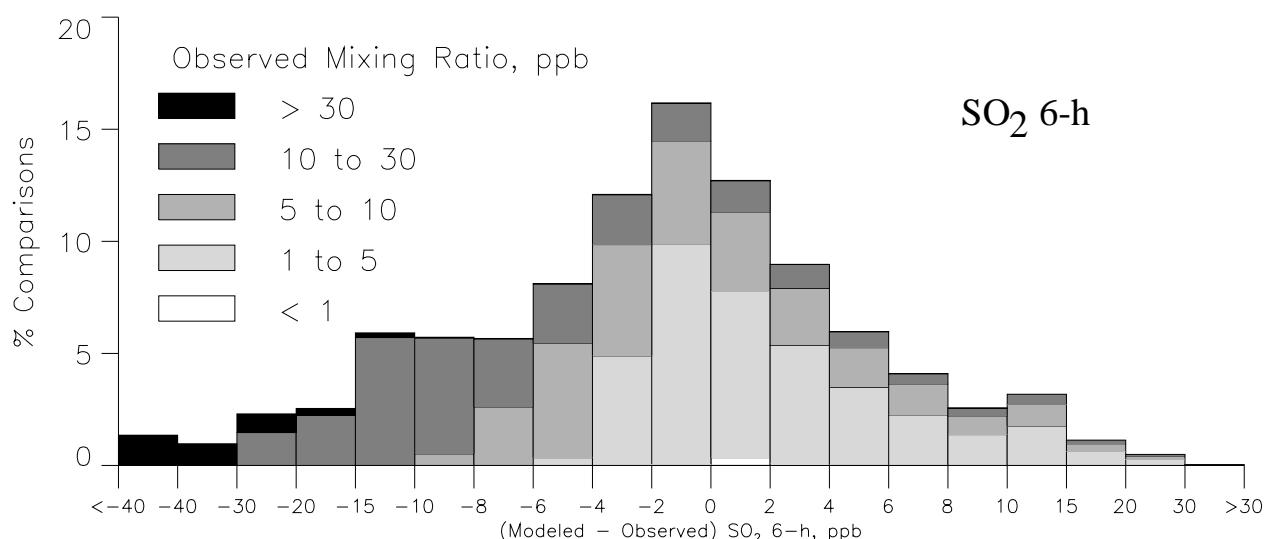
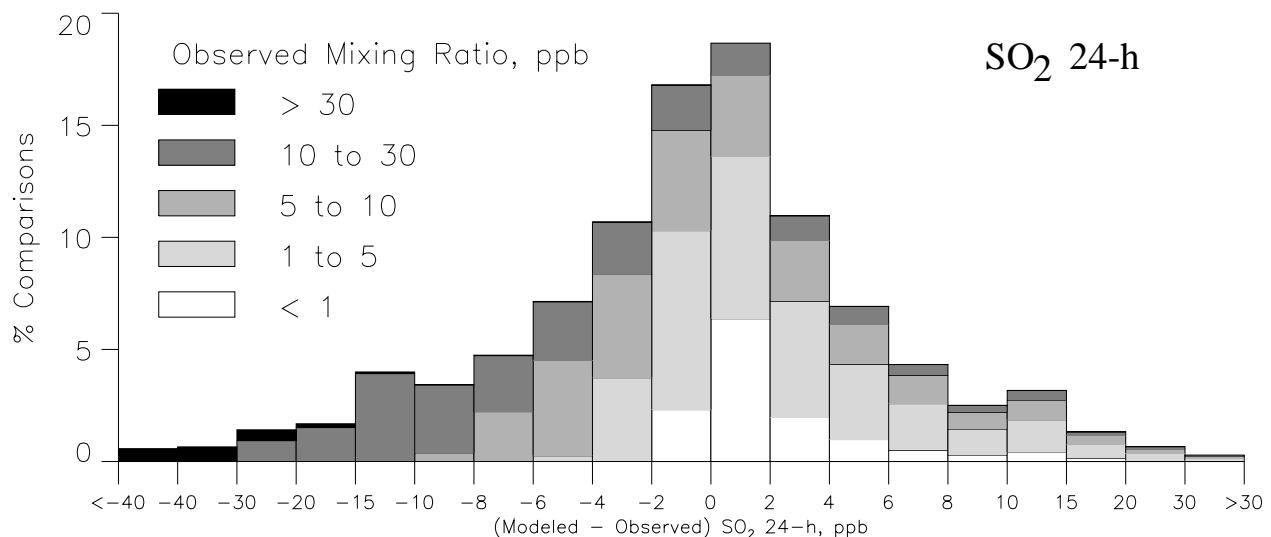
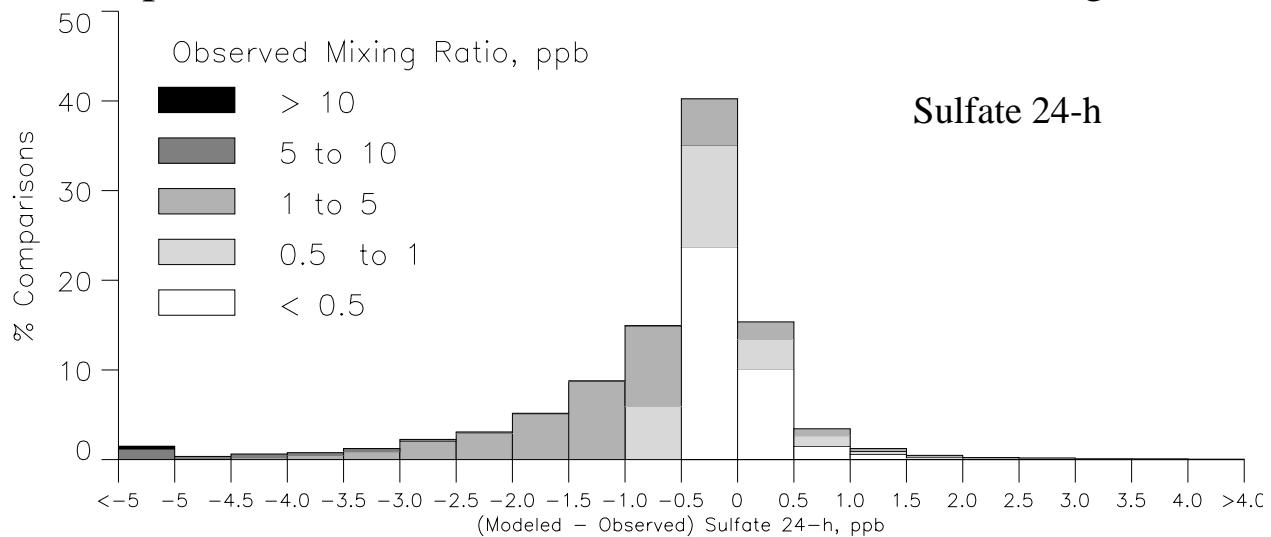
#Benkovitz & Schwartz, JGR 1997.

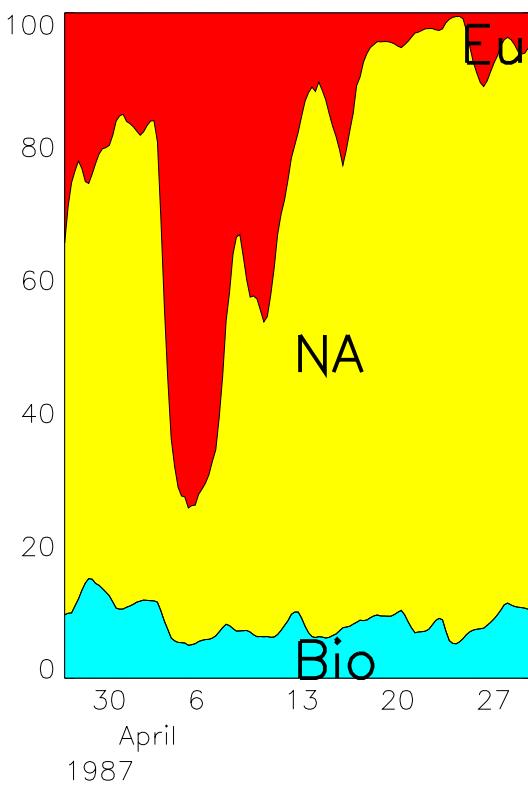
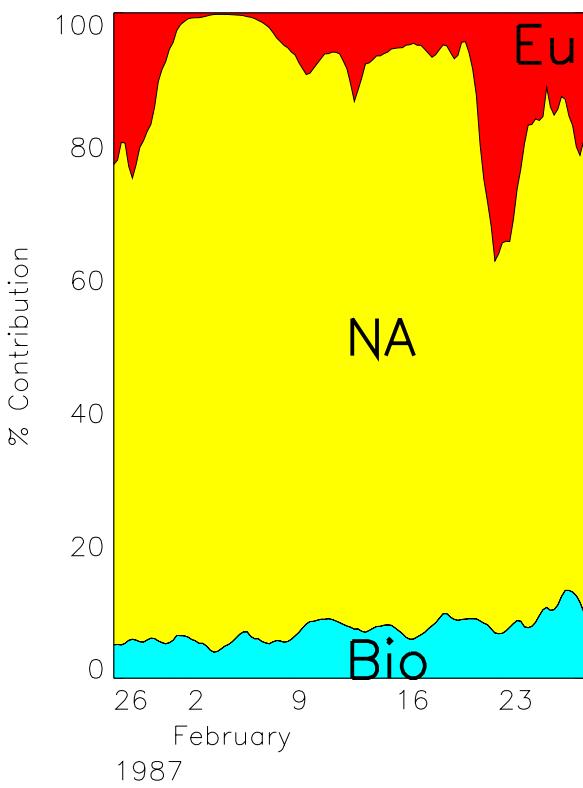
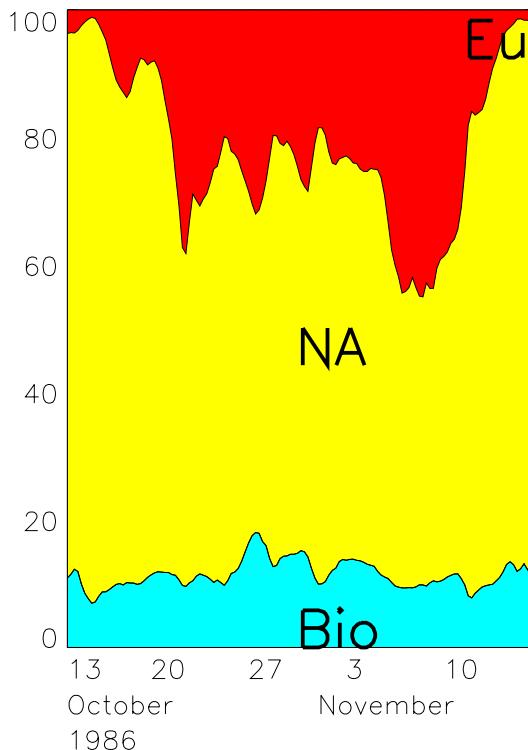
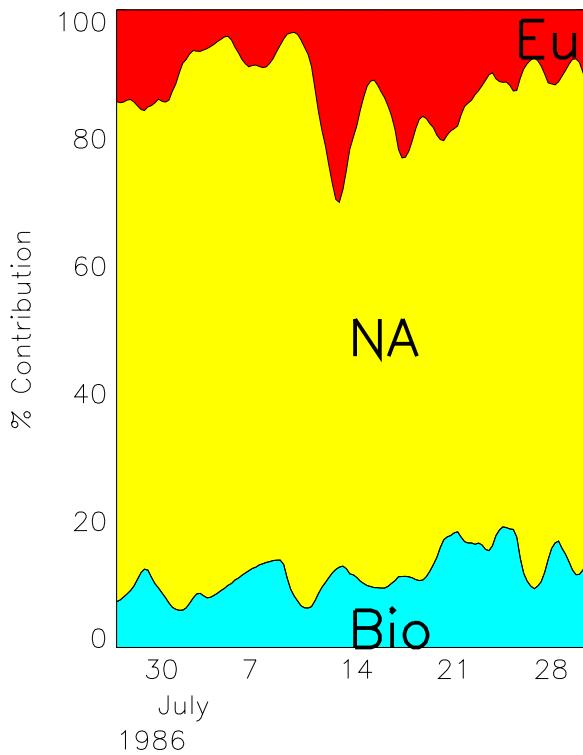
#Barth et al., JGR (submitted).

#Rasch et al., JGR (submitted).

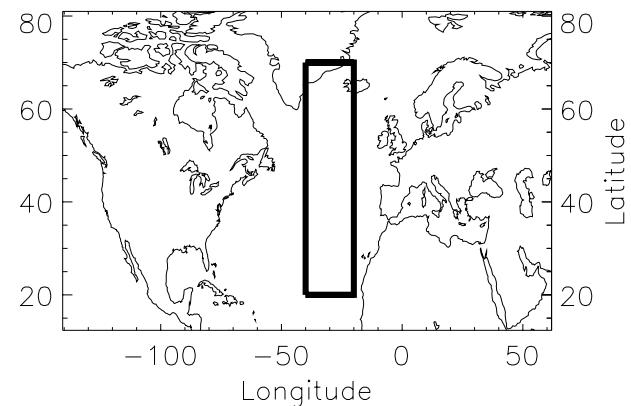
#Benkovitz et al., (in preparation).

Comparison of Modeled and Observed Surface Mixing Ratios



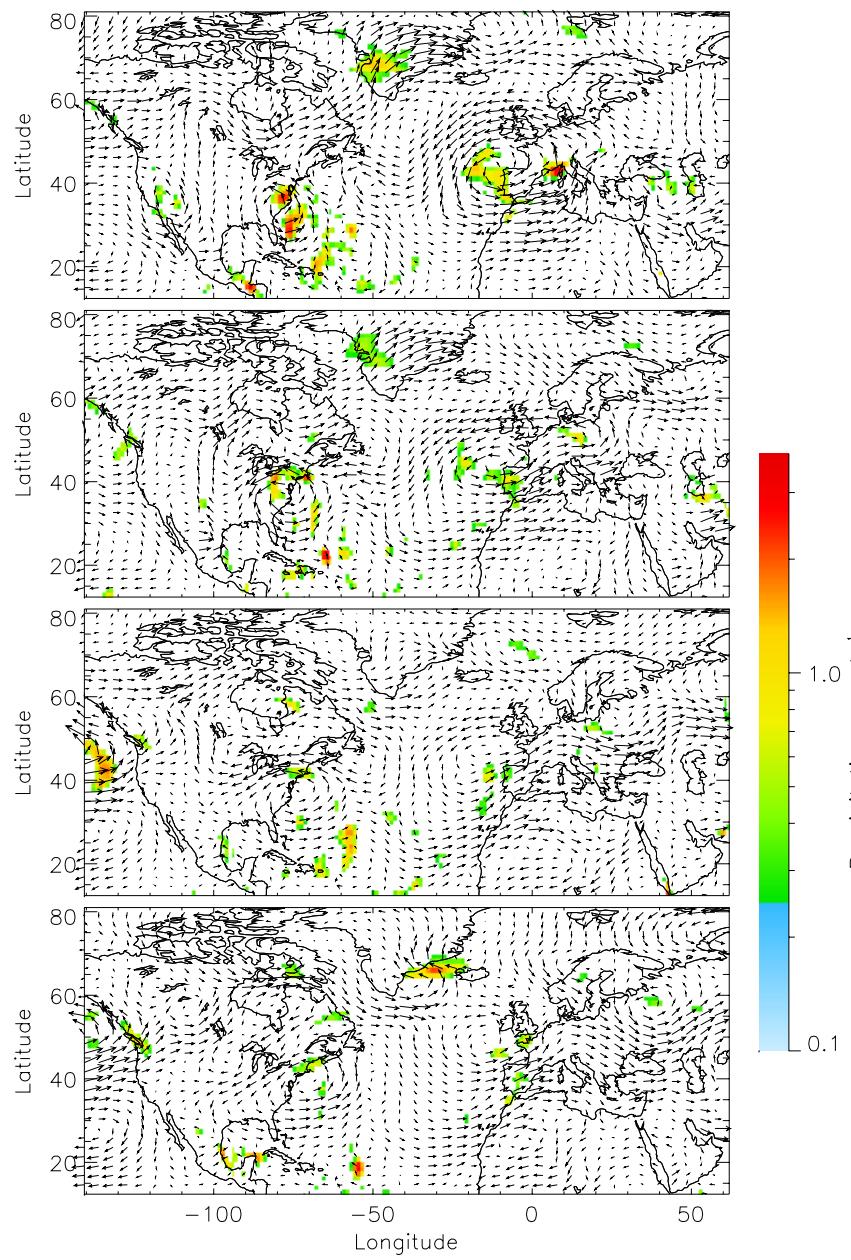


Contribution of Sources to Sulfate Burden



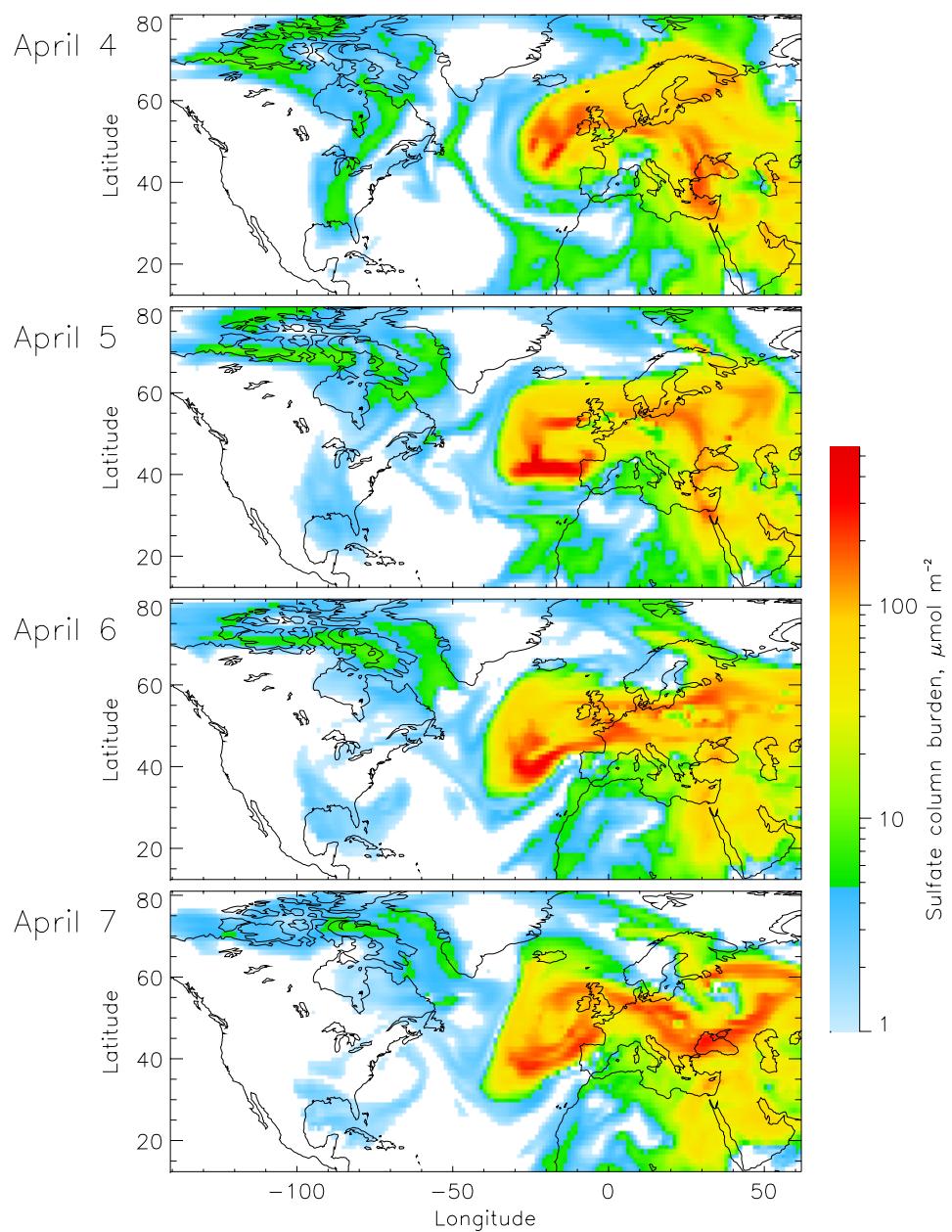
Bio=biogenic sources
 NA=anthropogenic sources
 west of 30W
 Eu=anthropogenic sources
 east of 30W

Evolution of the European Contribution to Sulfate Column Burden April 4 to 7, 1987



Max wind speed: 33.00 m s^{-1}

Winds at $\sim 850 \text{ hPa}$.



Sulfate column burden, $\mu\text{mol m}^{-2}$

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★ Meteorology

#ECMWF

! Initialized analysis surface and upper air data.

! Surface forecast data (precipitation and radiation).

$1^{\circ} \times 1^{\circ}$ horizontal resolution.

#27 vertical levels.

★ Transport

#Bott advection scheme as modified by Easter.

★ Dry deposition

#Time and location dependent using the methodology
of Wesely with the ECMWF meteorological data.

★ Wet deposition

#Berkowitz/Hale.

GChM-O Version 2

★ Chemistry

#Gas phase

- ! $\text{SO}_2 + \text{OH} \rightarrow \text{Sulfate}$
- ! $\text{DMS} + \text{OH} \rightarrow \text{Sulfate, MSA}$ (Yin/Seinfeld mechanism)
- ! $\text{HO}_2 + \text{HO}_2 \rightarrow \text{H}_2\text{O}_2$ (limited by MOZART mixing ratios)

#Aqueous phase (all clouds)

- ! $\text{SO}_2 + \text{H}_2\text{O}_2 \rightarrow \text{Sulfate}$
- ! $\text{SO}_2 + \text{O}_3 \rightarrow \text{Sulfate}$
- ! Cloud water pH estimated assuming sulfate is present as NH_4HSO_4 + the contribution of MSA.

#Mixing ratios of HO_2 , OH , H_2O_2 , and O_3 from MOZART model (Brasseur, JGR 103, 1998)

#Rate constants from NASA (1997)

Species Represented by the Model

Source ↓	SO ₂		Sulfate			DMS	MSA
Process →	Primary	Secondary Gas Phase	Primary	Secondary Gas Phase	Secondary Aqueous Phase	Primary	Secondary Gas Phase
Anthropogenic 140°W to 30°W	X		X	X	X		
Anthropogenic 30°W to 60°E	X		X	X	X		
Anthropogenic 60°E to 140°W	X		X	X	X		
Volcanic	X			X	X		
Biogenic		X		X	X	X	X
External		X		X	X		

Primary = directly emitted

Secondary = formed by atmospheric chemistry

GChM-O Version 2

Simulations

★ Temporal Domain

#November 1 to December 15, 1995 (ACE-1).
#June 1 to July 25, 1997 (ACE-2).

★ Geographic domain

#Southern Hemisphere to 81°S (ACE-1).
#Northern Hemisphere to 81°N (ACE-2).

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Emissions

★ Anthropogenic SO₂

- ! EDGAR 1990 inventory, Olivier et al., 1995.
- ! Regional emissions for Australia (F. Carnovale) and New Zealand (H. Plume, personal communication).

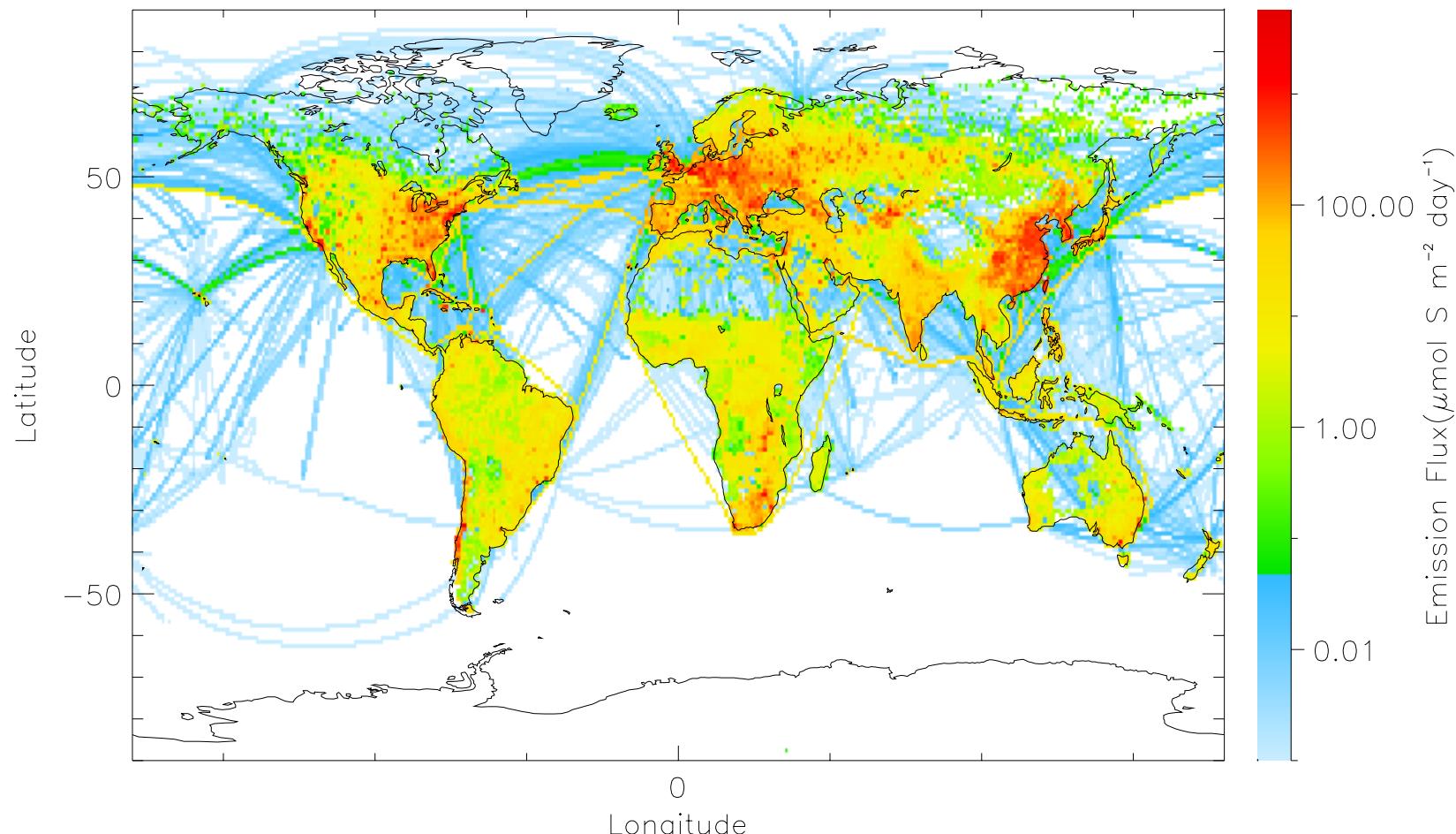
★ Biogenic DMS

- ! Oceanic: Kriging ocean measurements from Kettle et al., 1996. Kriging ACE-1 measurements to improve values in the ACE-1 experimental region. Kriging ACE-2 measurements to improve values in the ACE-2 experimental region.
- ! Land: B. Lamb (personal communication).

★ Volcanic SO₂ compiled at BNL.

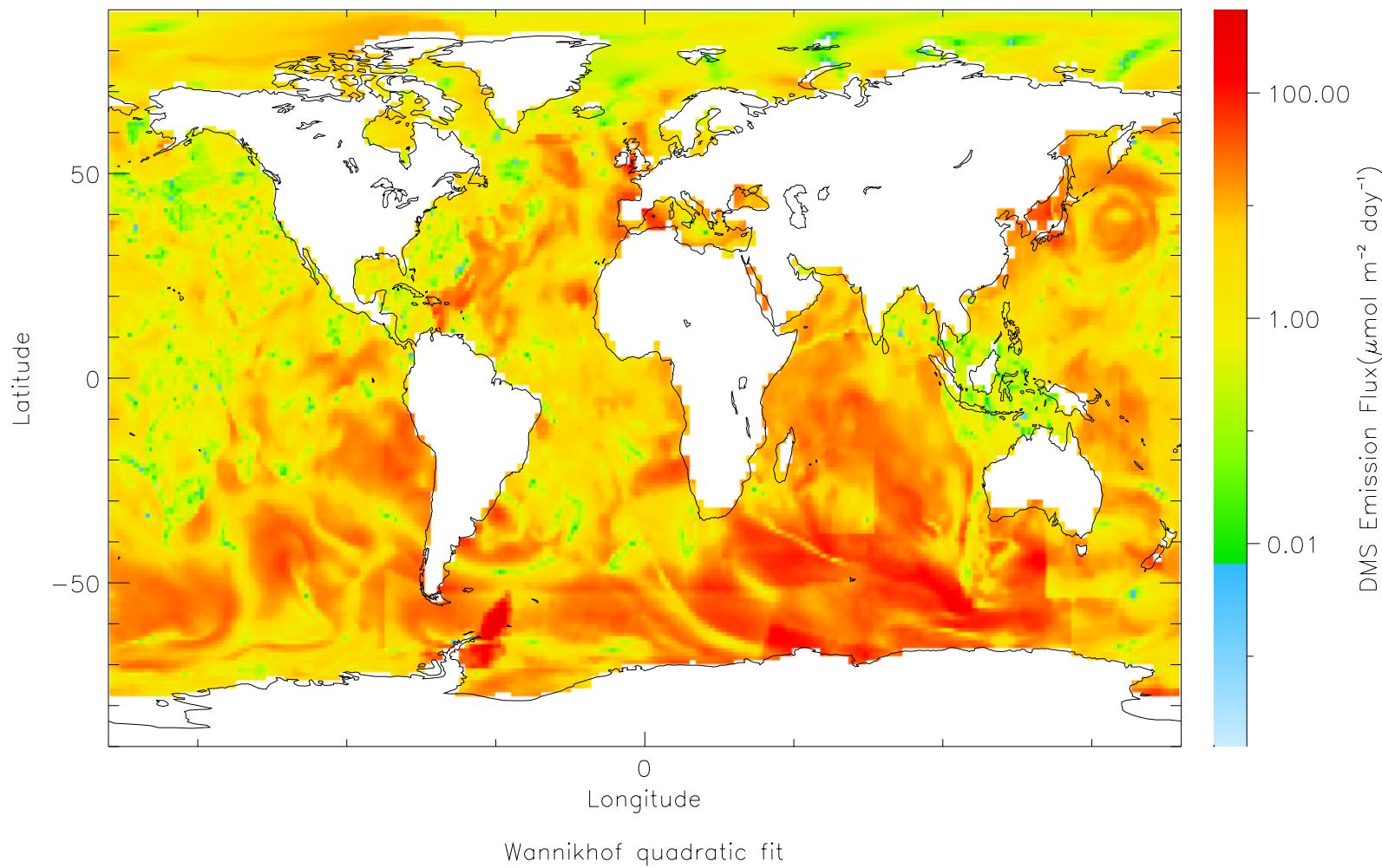
- ! Daily for November 1 to December 15, 1995.
- ! Daily for June 1 to July 25, 1997.

Anthropogenic Emissions of S for 1990

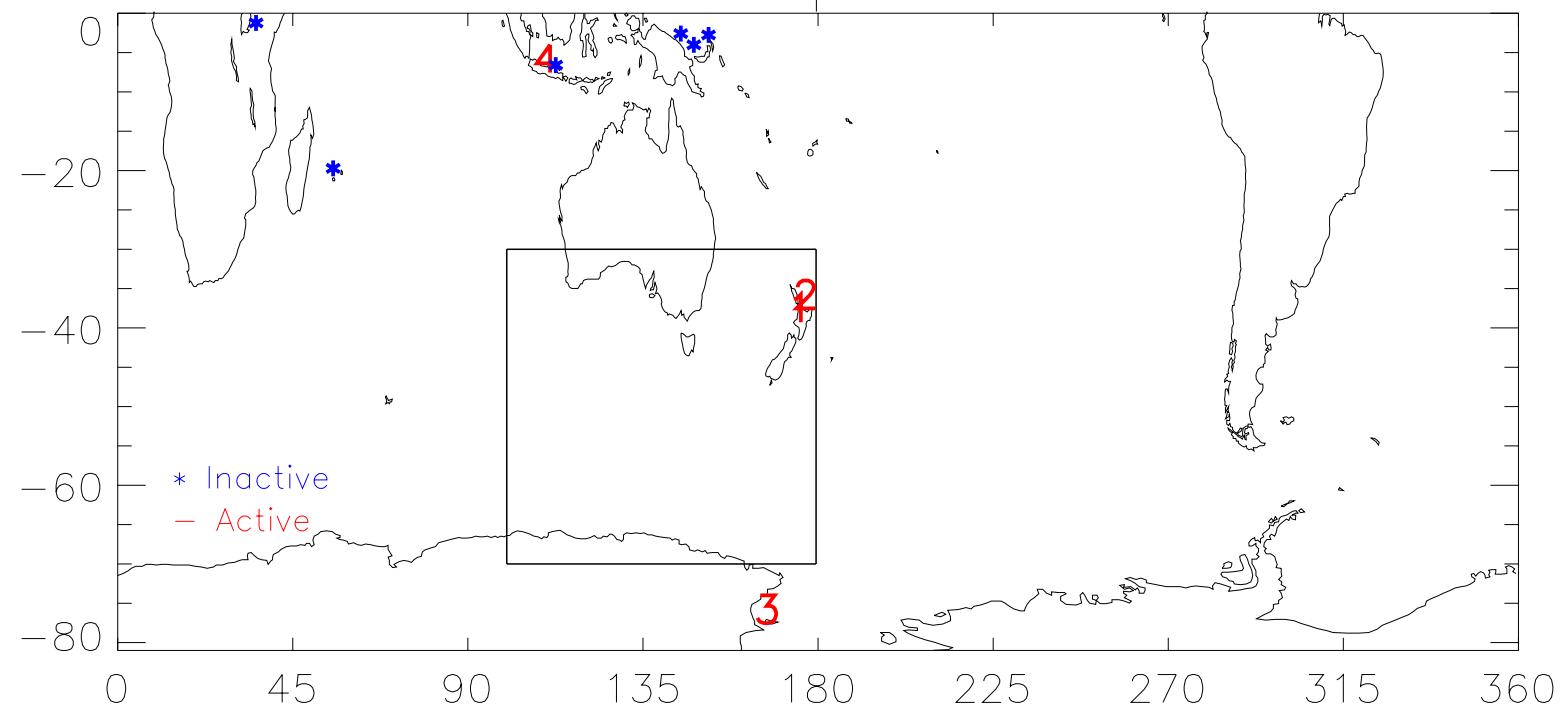


EDGAR Version 2.0

DMS Oceanic Emissions for 19951101



Southern Hemisphere Volcanoes



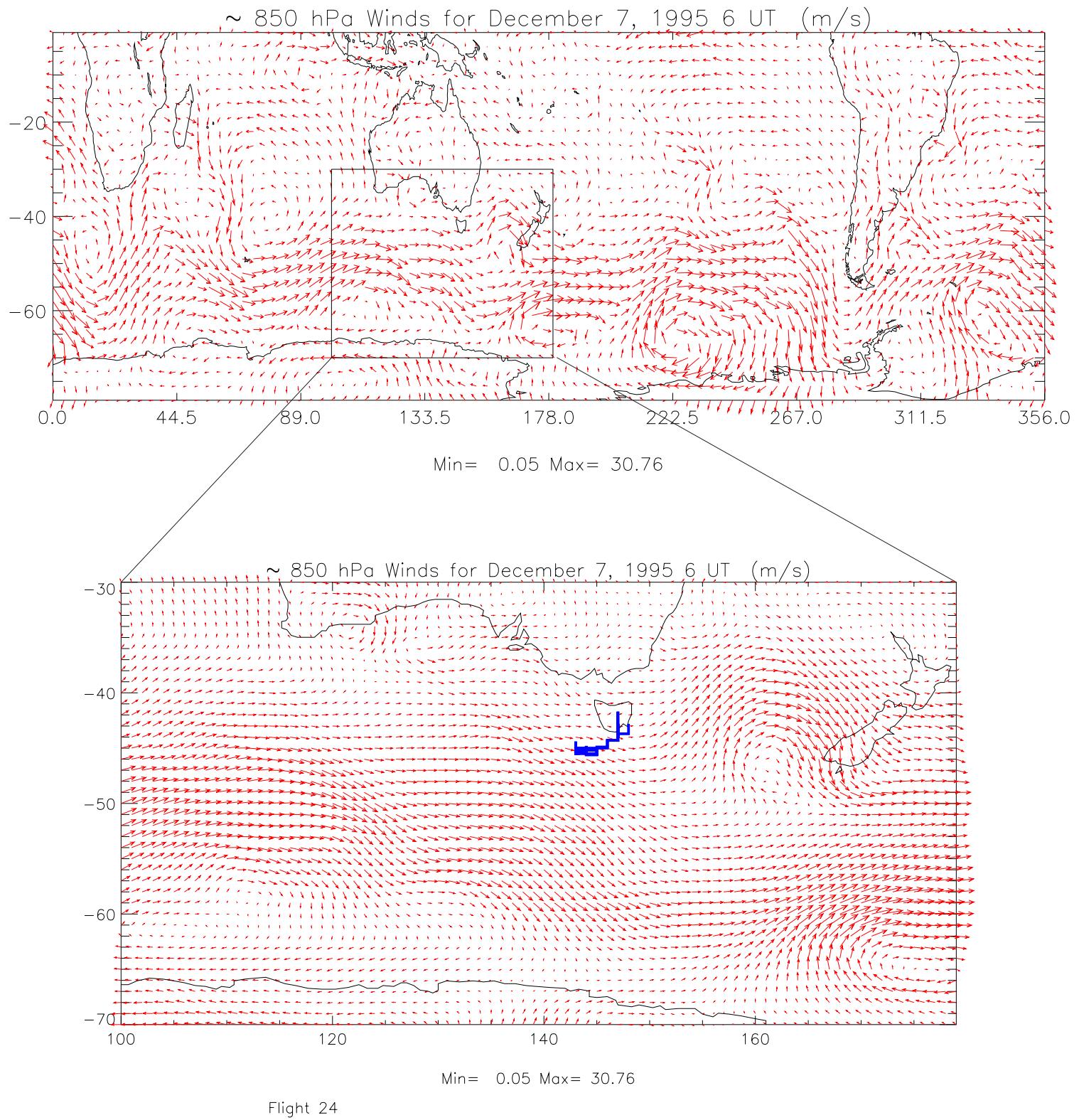
Emissions for Nov 1 to Dec 31, 1995
Name 10^6 g SO₂

1	Ruapehu	111010
2	White Island	19093
3	Erebus	3355
4	Merapi	5020

ACE-1

Location of Flight 24

December 7, 1995



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Incorporation of Aerosol Microphysical Processes

★ Quadrature Method of Moments (QMOM)

#Full Shugard- Heist-Reiss (SHR) model for binary nucleation of H_2SO_4 -water aerosol.

#Aerosol growth laws of arbitrary form.

#Closure of aerosol moment evolution equation via QMOM.

★ Aerosol dynamics

#Supplementation of QMOM with Lagrangian interpolation method.

#Multicomponent nucleation of NH_3 - H_2SO_4 -water.

#Coagulation dynamics.

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Incorporation of Aerosol Microphysical Processes

★ Coupled aerosol and cloud processes

#Explicit treatment of the size of cloud droplets
pertinent to indirect forcing.

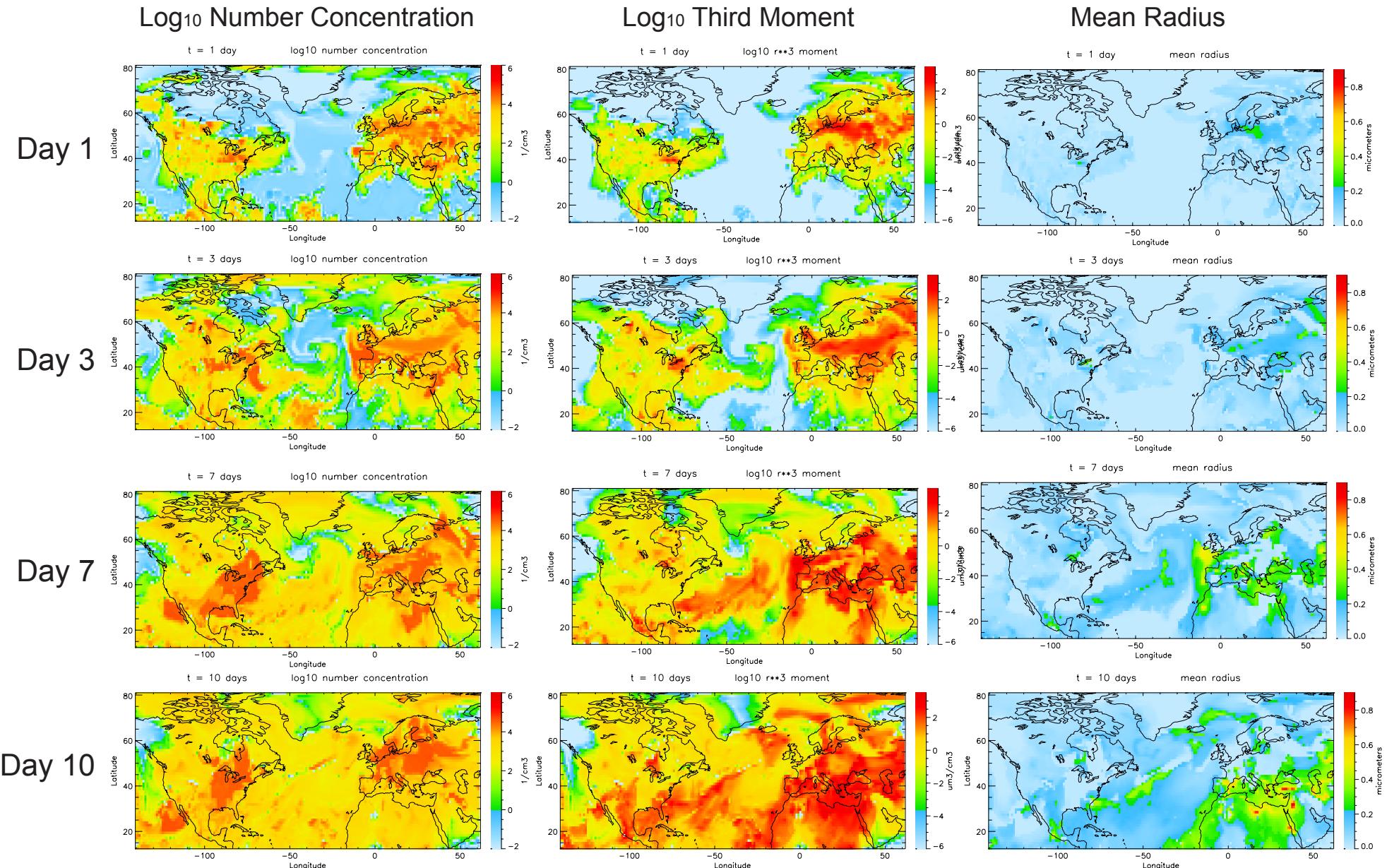
#Model will now include two coupled sets of six
moments each: one set for aerosol and one set for
the cloud distribution.

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Ten-Day Simulation Includes

- ★ Sulfate and sea salt aerosols.
- ★ Water uptake and release with changing RH.
- ★ Nucleation of H_2SO_4 /water aerosol via Jaecker-Voirol & Mirabel (1989).
- ★ Sea salt source via Smith et al. (1993).
- ★ Condensational growth.
- ★ Coagulation.
- ★ Size-resolved dry deposition.
- ★ Wet removal in precipitating clouds.

Results of a Ten Day Simulation



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Retrieval of Aerosol Optical Properties From Moments of The Particle Size Distribution

- ★ **The Multiple Isomomental Distribution Aerosol Surrogate (MIDAS) Technique**
 - #Retrieves families of isomomental surrogate distributions for the true size distribution from its lower moments.
 - #Each distribution is a multimodal lognormal or modified gamma distribution.
 - #For retrieved lognormals, the MIDAS technique 'broadens' the abscissas obtained from Gaussian quadrature into 'modes' of finite width.
 - #Optical properties are computed from an average of the retrieved distributions.

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Reduced forcing per unit volume ($\mu\text{m}^2 \text{ cm}^{-3}$) is defined as:

$$\Delta f_r = p \int r^2 Q_{sca}(r) b_u(r, q_o) f(r) dr$$

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Figure Caption.

Reduced forcing per unit volume ($\mu\text{m}^2 \text{ cm}^{-3}$) for each of the test distributions.

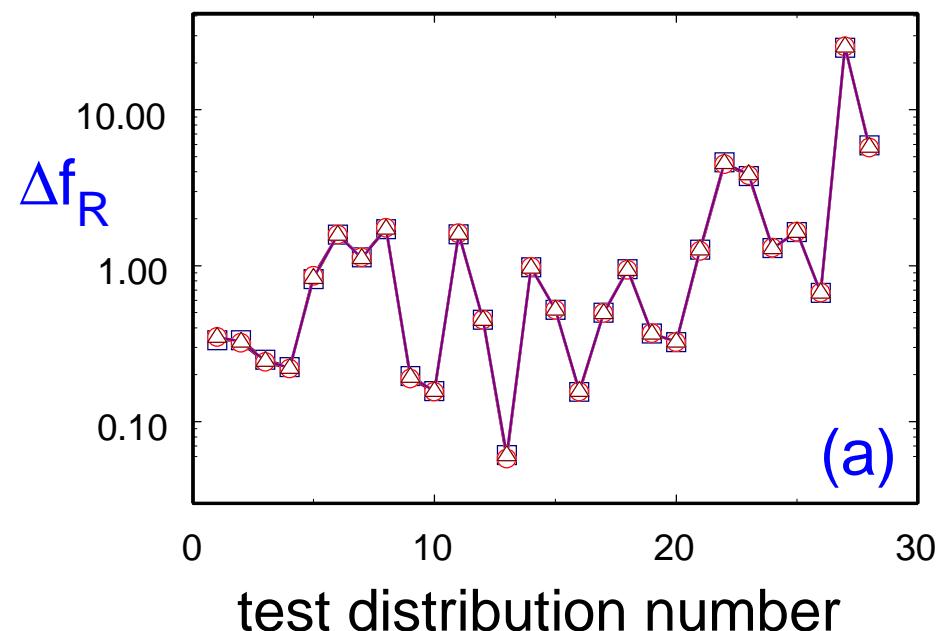
(a) Q = exact results, " = retrieved lognormal results,
 Δ = retrieved modified gamma results, at
 $\lambda = 632.8 \text{ nm}$ and $n = 1.55 - 0.1i$.

(b) Ratio of retrieved result to exact result.
" = lognormals, Δ = modified gammas,
= modified gammas averaged over the solar spectrum of *Coakley et al. (1983)*.

(c) Q = exact results, " = 3-point quadrature.

(d) Ratio of the quadrature result to the exact result.
" = the single wavelength results,
= the results averaged over the solar spectrum.

RETRIEVAL TECHNIQUE



3-POINT QUADRATURE

